

CHAPTER 11 CHEMICAL REACTIONS PRACTICE PROBLEMS ANSWERS



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Chapter 11. Chemical Equations 11.1. ChemicalEquation Concepts Under the right circumstances substances change into newsubstances. Suchchanges are calledchemical transformations,orreactions.The original substances are calledreac- tants,the substances resulting from the change are calledproducts.Chemical reactions reveal the basic properties of substances that form the heart of chemistry.

Chapter 11. Chemical Equations - University of New Mexico

Chapter 11: Chemical Reactions SC2 Students will relate how the Law of Conservation of Matter is used to determine chemical composition in compounds and chemical reactions. SC2.a. Identify and balance the following types of chemical equations: Synthesis, Decomposition, Single Replacement, Double Replacement, and Combustion SC2.b.

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330 Chapter 11 11.2 Types of Chemical Reactions Often charcoal briquettes ... Combustion is one of the ?ve general types of chemical reactions. In this chapter, you will learn that if you can recognize a reaction as being a particular type, you may be able to predict the products of the reaction. Guide for Reading Key Concepts

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Distinguish between the single- and double-replacement reactions indicated below by completing each, identifying each as to type, and explaining, in words, what has occurred in each.

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CHAPTER 11 Connecting chemical reactions and equations209 Sample problem 11.2 Write a sentence that completely describes the following reaction: $\text{N}_2\text{O}_3(\text{g}) + \text{H}_2\text{O}(\text{l}) \rightarrow 2\text{HNO}_2(\text{aq})$ Gaseous dinitrogen trioxide reacts with water to produce a solution of nitrous acid. Revision questions 1.